DETI.2 Kick-Off Meeting September 13, 2012, HZDR

Growth and electronic properties of TiO₂–based thin films



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Outline

- History of development
- Motivation
- Overview of activities
- Basics of transparent conductive oxides
- Challenges and objectives
- Experimental approach
- Basics of magnetron sputtering
- Direct growth of TiO₂-based TCOs
- Two-step formation process
- Summary and Outlook
- Acknowledgements



History of development

CdO: bulk 1902 (Streintz), thin film 1907 (Badeker)

- CdO-based compounds:
 - best electrical properties
 - o optical transmittance is not optimal, toxic material
- **SnO₂**: first patent 1938 (Littleton), films on glass 1942 (McMaster)
 - SnO₂:F is dominating on the market (Asahi Glass)
 - o established technology, integration in large area glass production
 - o fluorine-containing precursors, high environmental footprint
- In₂O₃: first patent 1947 (Zunick)
 - $-\ln_2O_3$:Sn (ITO)
 - o standard technology (sputtering), the best material available
 - o expensive, metallic indium supply problems
- **ZnO:AI**: first publication 1971 (Wasa)
 - \circ early stages of commercialization, cost-efficient production
 - environmentall stability problems
- TiO₂:Nb: first publication 2005 (Furubayashi)

Review: Ingram et al, *J. Electroceram.* 13, 167 (2004) Furubayashi et al, *Appl. Phys. Lett.* 86, 252101 (2005)

History of development





Motivation







Transparent conductive oxides (TCOs): key materials in optoelectronics

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Overview of activities

Wilde, Cornelius (FWIN), Liedke, Vinnichenko

> Bilateral, BMWi/AiF, BMBF-TUBITAK

HZB, von Ardenne, GfE Fremat, Bosch, Solayer, Vaciontec, DTF, Heliatek, CreaPhys, Interpane, Euroglas, LIMO, METU, Bilkent Univ.



Neubert, Cornelius (FWIN), Vinnichenko

Bilateral, BMBF MatRessource (submitted)

FEP, DTF, Solayer, Vaciontec, LIMO, Euroglas, HZDR Innovation

Basics of transparent conductive oxides (TCOs)





Basics of transparent conductive oxides



- Material can be highly doped: native defects + extrinsic impurities
- Degenerate n-type semiconductors

Medvedeva in Transparent Electronics: From Synthesis to Applications, Wiley 2010

Basics of transparent conductive oxides



- Photon Energy $\leq \hbar \omega_p$: free electron absorption in the near IR
- Plasma frequency $\omega_p \le 1 \text{ eV}$







- Photon Energy≥E_G: interband transitions
- Burstein-Moss shift due to free charge carriers

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e N_e \mu_e = \frac{1}{e^2 N_e \tau}
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Sn-doped indium oxide (ITO): best understood TCO !?

What is the band gap of In_2O_3 ?

- Generally accepted: direct gap of E_G=3.75 eV, based on optical investigations Hamberg et al, PRB 30, 299 (1984)
- X-ray photoelectron spectroscopy: fundamental gap is <3 eV
- In some cases optical data show absorption outset at 2.6 eV
- It is attributed to indirect transitions
 Christou, et al, JAP 88, 5180 (2000)
 McGuinness et al, PRB 68, 165104 (2003)
- Recent comprehensive study: fundamental band gap is not higher than 2.9 eV, direct gap
- But corresponding transitions are parity forbidden!
- Only transitions from lower levels (green) are allowed



Walsh et al, PRL 100, 167402 (2008)



Basics of transparent conductive oxides

Exclusively post-transition metal oxides?

- CdO, In₂O₃, SnO₂, ZnO-based:
 - $-(n-1)d^{10}ns^2$ electronic configurations
 - -four- (ZnO) or six-fold (CdO, In_2O_3 , SnO₂) coordinated atoms
 - Conduction band bottom: metal s +oxygen p
- TiO₂-based: exception



Basics of transparent conductive oxides



Not true for TiO₂-based TCOs!



Basics of transparent conductive oxides: TiO₂:Nb/Ta

- Difference to conventional transparent conducting oxides (TCOs):
 - Conduction band bottom:
 - no extended metal s-states
 - mainly formed by Ti 3d orbitals
 - Different phases/polymorphs different electrical properties:
 - Amorphous: non-conductive
 - Anatase: very well conductive (low m*)
 - Rutile: only weak conductivity (high m*)
 - High refractive index
 - By now highest quality films are achievable only using epitaxial substrates
- Challenges:
 - introducing controllable oxygen deficiency into the material
 - control of the phase composition
 - achieving high quality polycrystalline films

Hitosugi et al, Phys.Stat.Sol (a) 207, 1529 (2010)



Anatase



http://cst-www.nrl.navy.mil

Basics of transparent conductive oxides: TiO₂:Nb/Ta



FIG. 4. Calculated local density of states (DOS) for \Box_0^0 , Ti_i^0 , V_{Ti}^0 , and Nb_{Ti}^0 using GGA. The hosts VBM and CBM are indicated by vertical-dashed lines.

Osorio-Guillen, Lany, Zunger, Phys.Rev.Lett. 100, 036601 (2008)



Basics of transparent conductive oxides: TiO₂:Nb/Ta



Challenges and Objectives

"... understanding of the basic physics and chemistry of metal oxide surfaces lags a decade or more behind that of metals and semiconductors..."

Henrich and Cox, The Surface Science of Metal Oxides, Cambridge University Press, 2000

 Understanding even of the properties of metal oxide films and single crystals often lags behind

Basic understanding:

- Defect formation
- Incorporation and electrical activation of doping impurity
- Formation of secondary phases and their properties
- Limits to the charge carrier transport

Controllability and reproducibility:

- Fine tuning of the metal/oxygen flux
- Understanding role of the plasma energetic species
- Thin film microstructure and surface morphology

Addressing these issues is crucial to ensure further progress

Experimental approach: film synthesis



- Four sputter deposition systems
- DC and pulsed magnetron sputtering
 - Reactive (Ar + O₂), metallic (elemental and alloy) targets
 - Non-reactive (Ar), conductive (reduced) ceramic targets
- 2- and 3-inch magnetrons, balanced and unbalanced
- Energy-resolved mass spectrometry for detection of plasma ion species
- Langmuir probe with fine time resolution
- Precise tuning of metal/oxygen flux ratio:
 - Zn, Zn-Al, Zn-Ga: magnetron voltage
 - Ti, Ti-Nb: magnetron current
 - Reduced ceramic targets: optical emission of plasma intensity as a feedback

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Experimental approach: film characterization



- Annealing chamber (vacuum, Ar, O₂), equipped with in situ four point probe for resistivity measurements
- Annealing unit is transferrable to other experimental set-up: e.g. in situ XANES
- In situ spectroscopic ellipsometers: M-2000V, M-2000FI, J.A. Woollam Inc.
- Spectral photometer: SoliSpec 3700, Shimadzu



Basics of magnetron sputtering



 Energetic particle bombardment during growth may be used as a tool to affect the film structure and properties



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Ellmer, JPD 33, R17 (2000)

Role of energetic particles



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ZnO:Al

Metallic vs ceramic targets: plasma



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Direct growth, reactive magnetron sputtering, variation of deposition temperature



Reactive sputtering: hysteresis



pressure-voltage vs. flow hysteresis of Ti in (Ar,O₂), P=500W



Local atom arrangements



XANES data

- Ti2p spectra: 2p to 3d transitions
- L_{3,2} edges: spin-orbit splitting of Ti-2p core level into 2p_{3/2} and 2p_{1/2}
- XANES shows differences between as-grown disordered anatase- and rutile-like films, which are indistinguishable by XRD

Gago, Vinnichenko et al. Plasma Process. Polym. 7, 813 (2010)





Two-step process, DC magnetron sputtering of reduced ceramic targets



vacuum annealing at ~400°C, 1 hour



deposition of amorphous film (no heating) reduced ceramic target



Local atom arrangements







- Amorphous anatase-like films readily crystallize in good polycrystalline anatase TiO₂
- Disordered nc-rutile TiO₂ films remain stable upon annealing

Gago, Redondo-Cubero, Vinnichenko, Vazquez, Chem. Phys. Lett. 511, 367 (2011)



Relation to the structure and properties



2: the same films after annealing in vacuum



55

60

TiO₂:Nb

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Vinnichenko et al (in preparation)

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TiO₂:Ta: structure and properties



As-deposited

- insulating
- XRD-amorph.
- TEM-amorph.





Neubert et al (in preparation)

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TiO₂:Ta: optical properties, optimized films



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- The polycrystalline anatase TiO₂-based films with properties above current state of the art are realized by magnetron sputtering
- High-precision control of the oxygen deficiency and prevention of the rutile seed layer formation are crucial



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