# Antisite pair recombination in SiC by a modified concerted exchange mechanism

#### related publication:

Posselt, M., Gao, F., Weber, W. J.,

Atomistic simulations on the thermal stability of the antisite pair in 3C- and 4H-SiC, **Phys. Rev. B 73 (2006) 125206** 



# **Motivation**

# **D**<sub>I</sub> PL center:

- is observed at low temperature PL measurements, after electron, ion, or neutron irradiation, is found in as-grown material and in epitaxial layers, in *n* and *p* type material, in *3C*, *4H*, *6H* and *15R* polytypes
- is found to be stable after thermal treatment to above 2000 K

D <sub>I</sub> (zero phonon) lines	band gap
<i>3C:</i> L <sub>1</sub> = 1.972 eV	3C: 2.36 eV
<i>4H:</i> L <sub>1</sub> = 2.901 eV	<i>4H</i> : 3.23 eV
<i>6H:</i> L <sub>1</sub> = 2.625 eV, L <sub>2</sub> = 2.600 eV, L <sub>3</sub> = 2.570 eV	<i>6H</i> : 3.00 eV



#### interpretation



- native defect acts as a hole trap
- strongly localized hole captures an electron
- bound exciton is formed
- recombination between hole and electron gives the  $L_1$  emission line



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# candidates for the native defect:

Eberlein et al. 2002: Gali et al. 2003: Gali et al. 2003:

Pinheiro, Rauls et al. 2004/2005: Si<sub>c</sub>(C<sub>Si</sub>)<sub>2</sub> complex

#### however:

calculation of spectra is based on the existence of  $Si_{C}C_{Si}$ and  $Si_{C}(C_{Si})_{2}$ , their thermal stability has not yet been checked

# *this work:* dynamic simulations on the thermal stability of Si<sub>C</sub>C<sub>Si</sub> in 3C- and 4H-SiC







# **Simulation method**

#### systematic classical MD simulations

- interatomic potential of Gao et al.
- supercell with 1152 (3C-SiC) or 1920 (4H-SiC) atoms

#### ab-initio MD simulations for selected examples

- DFT-code NWCHEM
- supercell with 128 atoms (3C-SiC)



#### determination of the thermal stability of Si<sub>c</sub>C<sub>Si</sub>:

- system with Si<sub>c</sub>C<sub>si</sub> is heated rapidly to a given temperature
- lifetime of  $Si_C C_{Si}$  and the atomic recombination mechanism are analyzed by different independent methods
- due to the statistical nature of the recombination process, several statistically independent simulations must be performed at each temperature



#### accuracy of the interatomic potential of Gao et al:

#### V and I formation energies and migration barriers agree reasonably well with results of ab-initio calculations

structure and energetics of the antisite pair are also in good agreement with data obtained by ab-initio methods



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#### antisite pair configurations in 3C- and 4H-SiC



4H-SiC

C and Si on cubic sites (1), mixed configuration (2)



	СР			SCC-DFTB	DF					
Reference	this work		Rauls et al. 2001	Wang et	Torpo et	Eberlein et al. 2002		Gali et	Gali <i>et al</i> .	
					<i>al.</i> 1988	al. 1999			al. 2003	2003
	3C-SiC	4H-SiC,	4H-SiC,	4H-SiC, c,	3C-SiC	3C-SiC	4H-SiC, c	4H-SiC,	3C-SiC	4H-SiC
		atoms on two	other	atoms on two				not 🛛 c		
		hexagonal sites	configurations	cubic sites						
formation energy (eV)	6.65	6.72	6.66	5.2	5.9	5.8	5.68	5.33	6.1	5.9
binding energy (eV)	2.82	2.75	2.83	2.5	2.5	2.49	2.07	2.42		
Si-C bond length	-3%	-4%	-3%	-2%		-2%				
C-C bond length	-10%	-10%	-10%	-12%		-12%				
Si-Si bond length	+24%	+27%	+24%	+15%		+16%				



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## **Results for 3C-SiC**

The antisite pair recombination was studied at eleven temperatures between 800 and 2500 K (10 statistically independent simulations at each temperature).







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#### time dependence of the recombination parameter (rp) and evolution of the bond length between the atoms of the antisite pair



#### antisite pair recombination consists of 4 stages



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In *3C-SiC* the thermal stability of the antisite pair is low. It can be therefore <u>not</u> correlated with the D<sub>1</sub> PL center that is stable up to 2000 K.



### modified concerted exchange mechanism



Si moves to a position close to a tetrahedral site



 $V_{C}$ +Si<sub>TC</sub>+C<sub>Si</sub>



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configuration changes slightly, but remains in a {110} plane



 $V_{C}$ +Si<sub>TC</sub>+C<sub>Si</sub>



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3

rotation of the Si –C pair in the {110} plane



bond defect, type 1

further rotation of the Si – C pair, gradual change of the rotation plane until another {110} plane is reached



bond defect, type 2



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rotation of the Si –C pair in the new {110} plane

#### perfect lattice



energetics of the modified concerted exchange





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# bond defect in Si *Tang et al.* result of incomplete V-I recombination 1997 2 2 3 all atoms are fourfold coordinated



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#### bond defects in 3C-SiC

# bond defect,

type 1



coordination of atoms: ~4 nearest neighbors of Si: 2 x Si, 2 x C nearest neighbors of C: 2 x Si, 2 x C bond defect

type 2



coordination of atoms: ~4 nearest neighbors of Si: 1 x Si, 3 x C nearest neighbors of C: 3 x Si, 1 x C





Similarly to Si, the two bond defects in SiC can be considered as the result of incomplete recombination of  $V_C$  and <100>C+Si. This process is therefore one of the possible pathways for antisite pair formation.



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## **Results for** *4H***-***SiC*

The antisite pair recombination was studied at nine temperatures between 900 and 2000 K.

C and Si on cubic sites (1), mixed configuration (2)



results are almost identical to those for *3C-SiC* 



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The lifetime of the antisite pair with atoms on hexagonal sites is different to that of the three other types of antisite pairs.

The recombination mechanism is <u>not</u> a concerted exchange. It is rather characterized by the separate motion of the atoms. Two reaction paths were identified:

- (i) main path: only the two atoms of the antisite pair move significantly.
- (ii) other path: motion of more than two atoms.

In *4H-SiC* the thermal stability of the four types of antisite pairs is low. They can be therefore <u>not</u> correlated with the D<sub>I</sub> PL center that is stable up to 2000 K.



# Summary

- The thermal stability of the antisite pair in *3C* and *4H*-*SiC* is low. Therefore, it is not a candidate for the D<sub>1</sub> PL center.
- The recombination of the antisite pair in *3C-SiC* and of three antisite pair configurations in *4H-SiC* occur via a modified concerted exchange, with two types of bond defects as intermediate states.



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